



Molecular Dynamics & Theoretical Chemistry

8 March 2013

Michael R. Berman

Program Officer
AFOSR/RTE

Air Force Office of Scientific Research

Integrity ★ Service ★ Excellence

Report Documentation Page				Form Approved OMB No. 0704-0188	
Public reporting burden for the collection of information is estimated to average 1 hour per response, including the time for reviewing instructions, searching existing data sources, gathering and maintaining the data needed, and completing and reviewing the collection of information. Send comments regarding this burden estimate or any other aspect of this collection of information, including suggestions for reducing this burden, to Washington Headquarters Services, Directorate for Information Operations and Reports, 1215 Jefferson Davis Highway, Suite 1204, Arlington VA 22202-4302. Respondents should be aware that notwithstanding any other provision of law, no person shall be subject to a penalty for failing to comply with a collection of information if it does not display a currently valid OMB control number.					
1. REPORT DATE 08 MAR 2013		2. REPORT TYPE		3. DATES COVERED 00-00-2013 to 00-00-2013	
4. TITLE AND SUBTITLE Molecular Dynamics and Theoretical Chemistry				5a. CONTRACT NUMBER	
				5b. GRANT NUMBER	
				5c. PROGRAM ELEMENT NUMBER	
6. AUTHOR(S)				5d. PROJECT NUMBER	
				5e. TASK NUMBER	
				5f. WORK UNIT NUMBER	
7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES) Air Force Office of Scientific Research ,AFOSR/RTE,875 N. Randolph,Arlington,VA,22203				8. PERFORMING ORGANIZATION REPORT NUMBER	
9. SPONSORING/MONITORING AGENCY NAME(S) AND ADDRESS(ES)				10. SPONSOR/MONITOR'S ACRONYM(S)	
				11. SPONSOR/MONITOR'S REPORT NUMBER(S)	
12. DISTRIBUTION/AVAILABILITY STATEMENT Approved for public release; distribution unlimited					
13. SUPPLEMENTARY NOTES Presented at the AFOSR Spring Review 2013, 4-8 March, Arlington, VA.					
14. ABSTRACT					
15. SUBJECT TERMS					
16. SECURITY CLASSIFICATION OF:			17. LIMITATION OF ABSTRACT Same as Report (SAR)	18. NUMBER OF PAGES 28	19a. NAME OF RESPONSIBLE PERSON
a. REPORT unclassified	b. ABSTRACT unclassified	c. THIS PAGE unclassified			



2013 AFOSR SPRING REVIEW 3002N PORTFOLIO OVERVIEW



NAME: Michael R. Berman

BRIEF DESCRIPTION OF PORTFOLIO:

Research on understanding and exploiting chemical reactivity and energy flow in molecules to improve Air Force systems, processes, and materials.

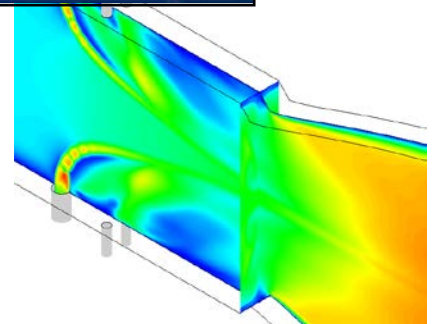
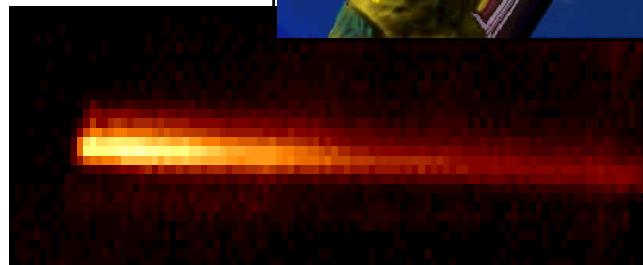
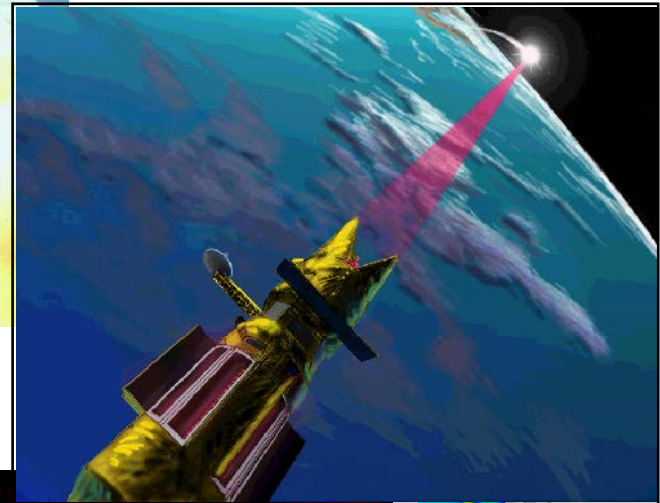
From a molecular perspective, understanding reaction mechanisms and developing predictive capabilities.

Understanding and utilizing chemical reactivity and catalysis for improved storage and utilization of energy

LIST SUB-AREAS IN PORTFOLIO:

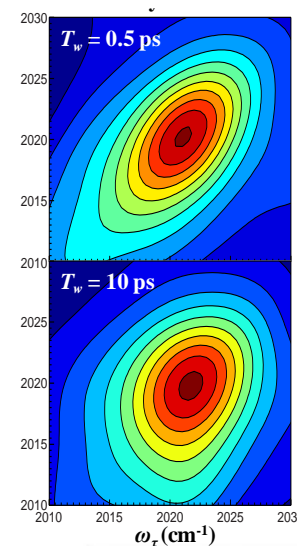
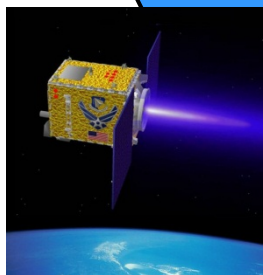
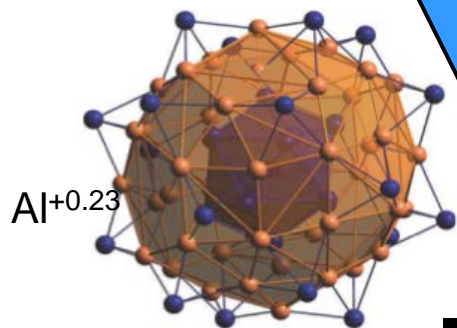
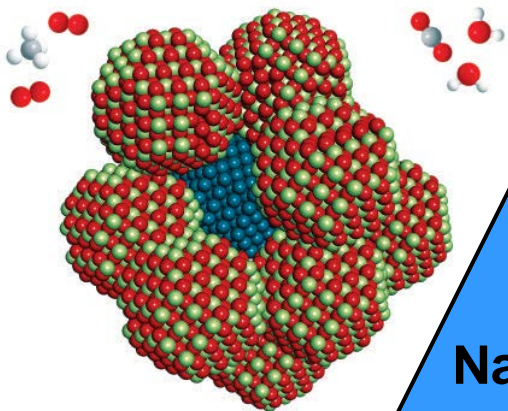
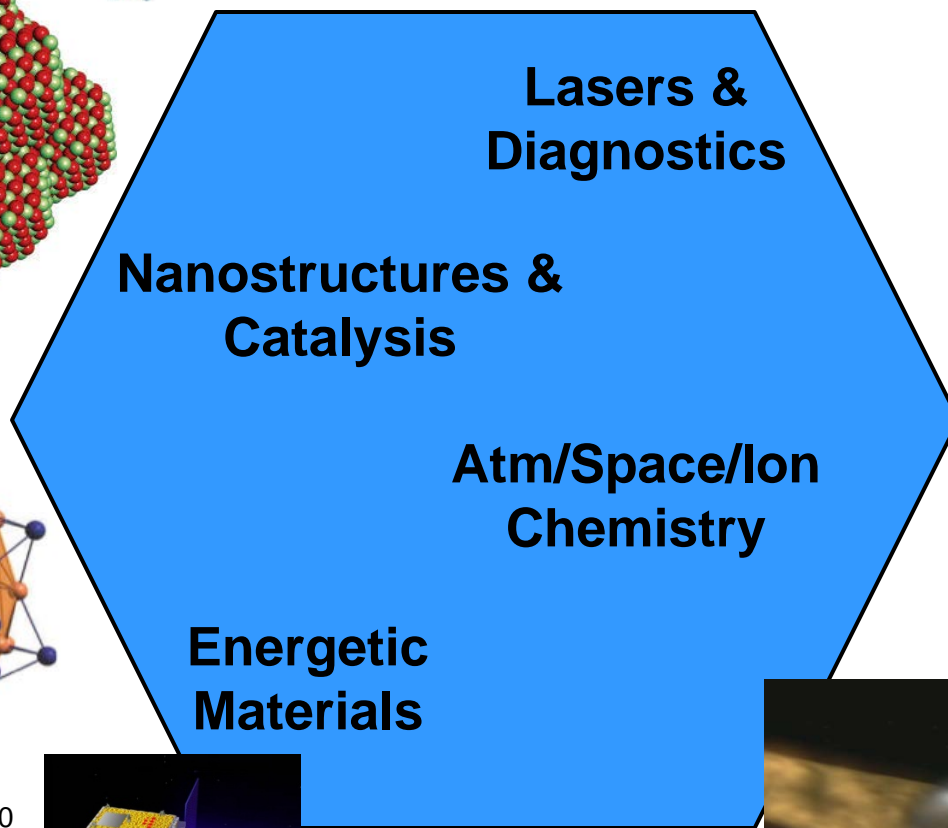
Molecular Dynamics

Theoretical Chemistry





Program Synergies

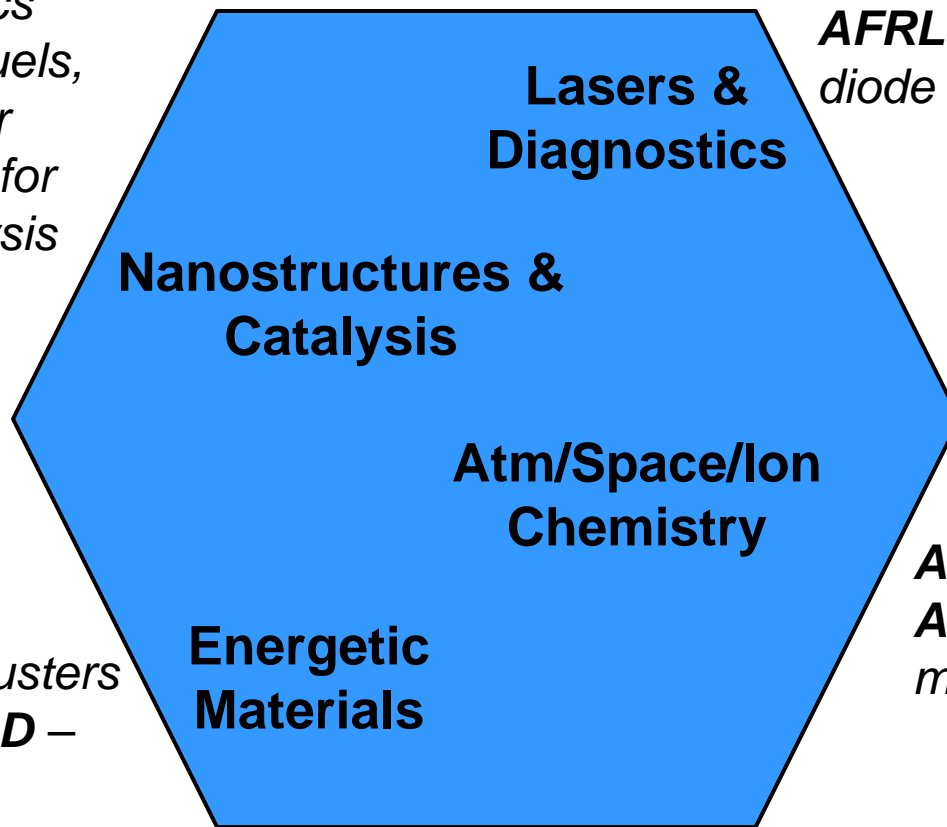




Program Interactions



ARO – plasmonics
AFOSR - Endo fuels,
combustion, solar
PNNL – Institute for
Integrated Catalysis



AFRL, AFOSR – XPAL, etc.
AFRL – ultrafast methods,
diode laser spect (ICOS)

Navy, DTRA – Clusters
AFRL, NASA, DoD –
Ionic Liquids

AFRL, PSI, SSI - Codes
AFOSR - Simulation
methods, Pls

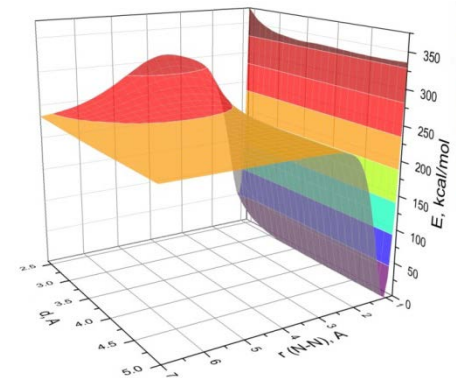
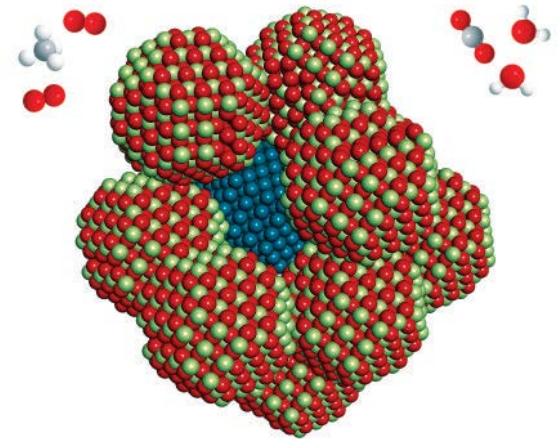
AFRL – RQ, RV, RX, RW
Quantum Chemistry Codes



Scientific Challenges



- **Storing energy in chemical bonds as fuels/propellants/munitions**
 - $\text{CO}_2 \rightarrow \text{JP8}$
- **Creating novel materials from nanoscale building blocks that can control energy flow**
 - Predict, prepare, probe
- **Predicting and controlling energy transfer in complex, reactive environments**
- **Probing processes at interfaces**

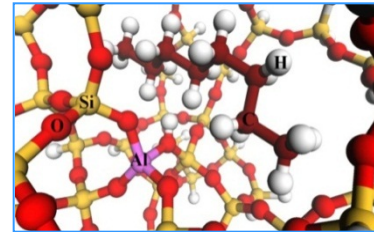




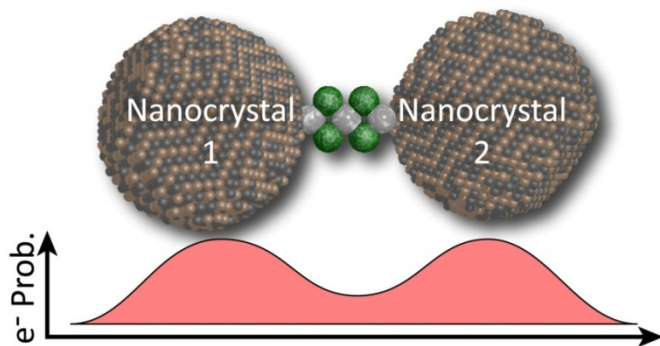
BRI Topics



- **Catalysis for Endothermic Fuels**
- **Foundations of Energy Transfer**



- **Plasma-Surface Interactions**
- **Nanoscale Building Blocks for Novel Materials**





Program Trends



- Catalysis
- Nanostructure Assemblies
- Plasma-Surface Interactions
- Ionic Liquid Propellants
- Real-time probing of reactions
- Hybrid Chemical Lasers
- Sensors for Trace Detection



AFRL Ionic Liquid Selected by NASA as Green Propellant

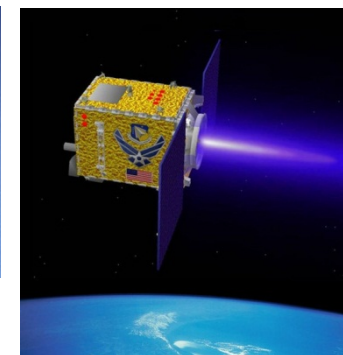


- *Ionic Liquid propellant developed by AFRL (AF-M315E; Hawkins et al) with support from AFOSR selected by NASA for "Green Propellant Infusion Mission"*
- *NASA will invest \$45M over 3 years*
- *Bridges gap between tech development and use*
- *Fly and characterize high performance green propellant in space in an integrated propulsion system- Falcon launch in 2015*
- *Team selected includes Ball Aerospace, Aerojet, AFRL, NASA*

AF-M315E



Monopropellant thruster



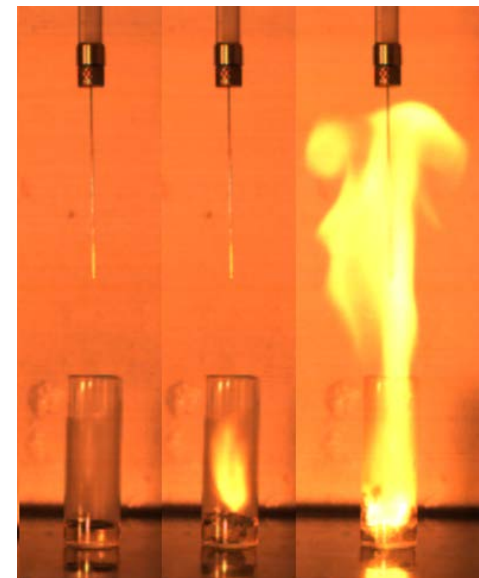
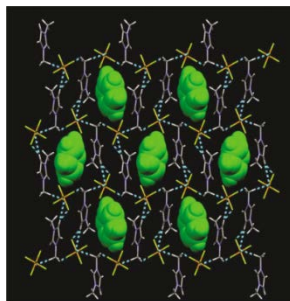
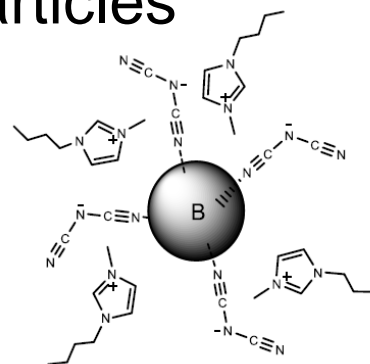
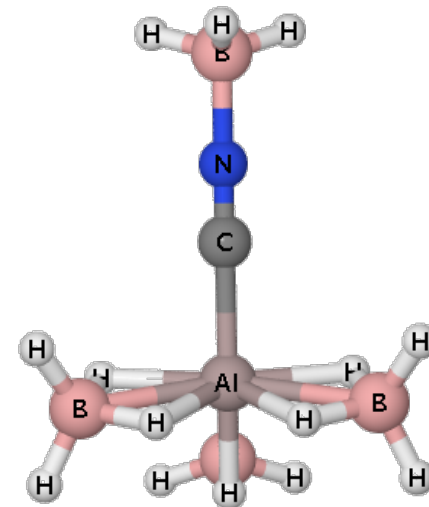
"A high performance green propellant has the potential to revolutionize how we travel to, from and in space" Michael Gazarik, Director of NASA's Space Technology Program.



New Ionic Liquids for Propellants



- Alumino-cyanoborohydride anions
 - Bipropellant fuel approach with high hydrogen content & large $\Delta H_{\text{combustion}}$
 - Hypergolic
- Additives to control energy and physical properties of ionic liquids
 - Graphenes greatly lower IL viscosity as internal 'lubricant' & increase reactivity
- Ionic liquids with metal nanoparticles
 - Passivating B and Al metal nps
- Liquid clathrates

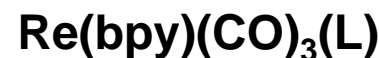
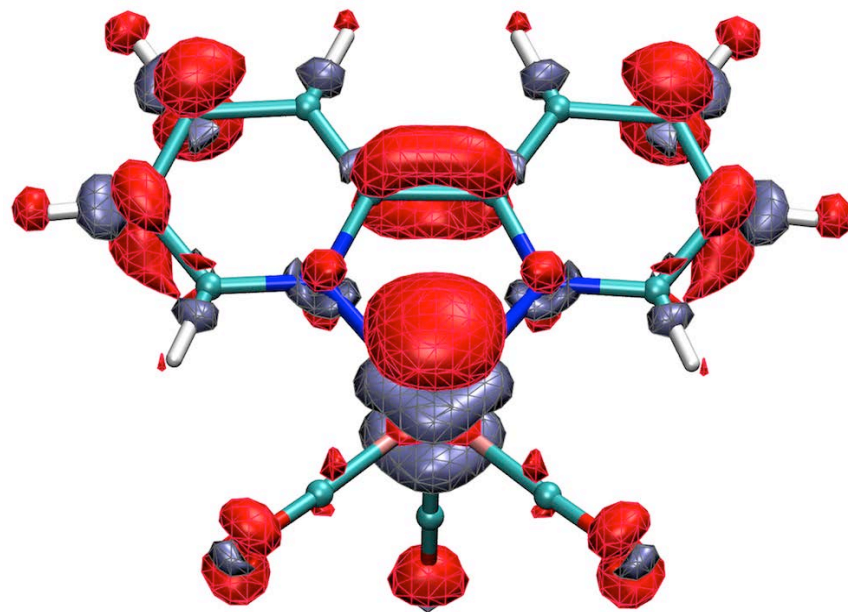




Mechanism of CO₂ Reduction



- Why is rhenium-based catalyst efficient at CO₂ reduction to CO?
 - hydride reduction is energetically favored, but CO₂ reduction dominates
- Make, model, measure approach
 - Synthesis
 - DFT studies
 - X-ray probing
- Oxidation state is Re⁰(bpy)⁻¹. Ligands are actively involved in bonding
- HOMO permits ***σ-bonding*** and ***π-bonding***; favors CO₂ binding
- Understanding mechanism permits ways to find alternative catalysts with similar electronic structure (Mn).



*Kubiak and co-workers, Angew. Chem., 2012
MURI*

*Kubiak, UCSD
Nilsson, Stanford
Carter, Princeton*

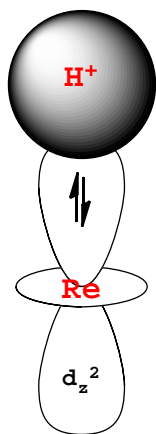




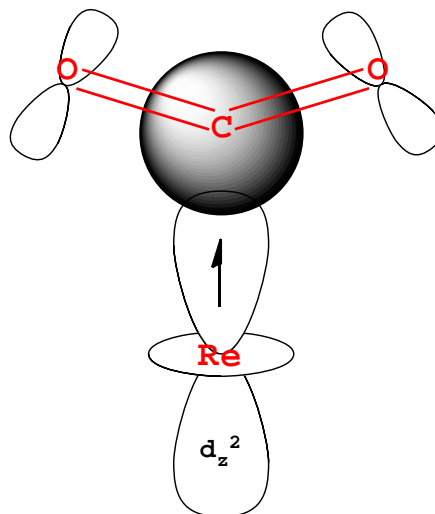
Proton Reduction vs. CO₂ Reduction



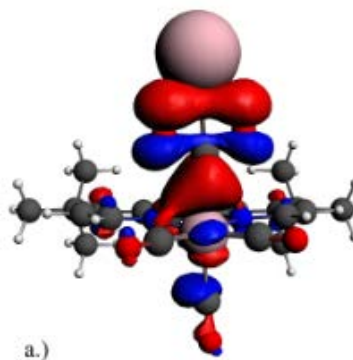
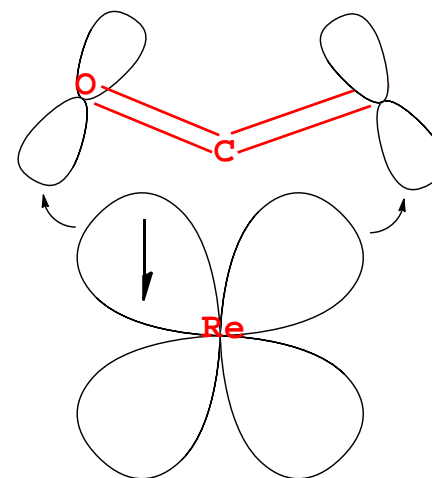
σ -bonding only



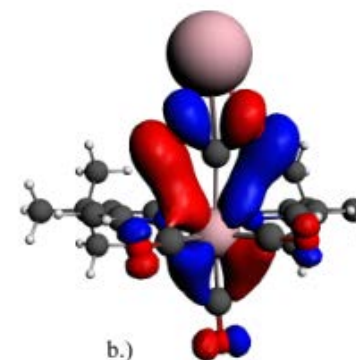
σ -bonding



π -bonding



a.)



b.)

Smieja, Benson, Kumar, Grice, Seu, Miller, Mayer, Kubiak ; *PNAS* **109**, 15646 (2012).

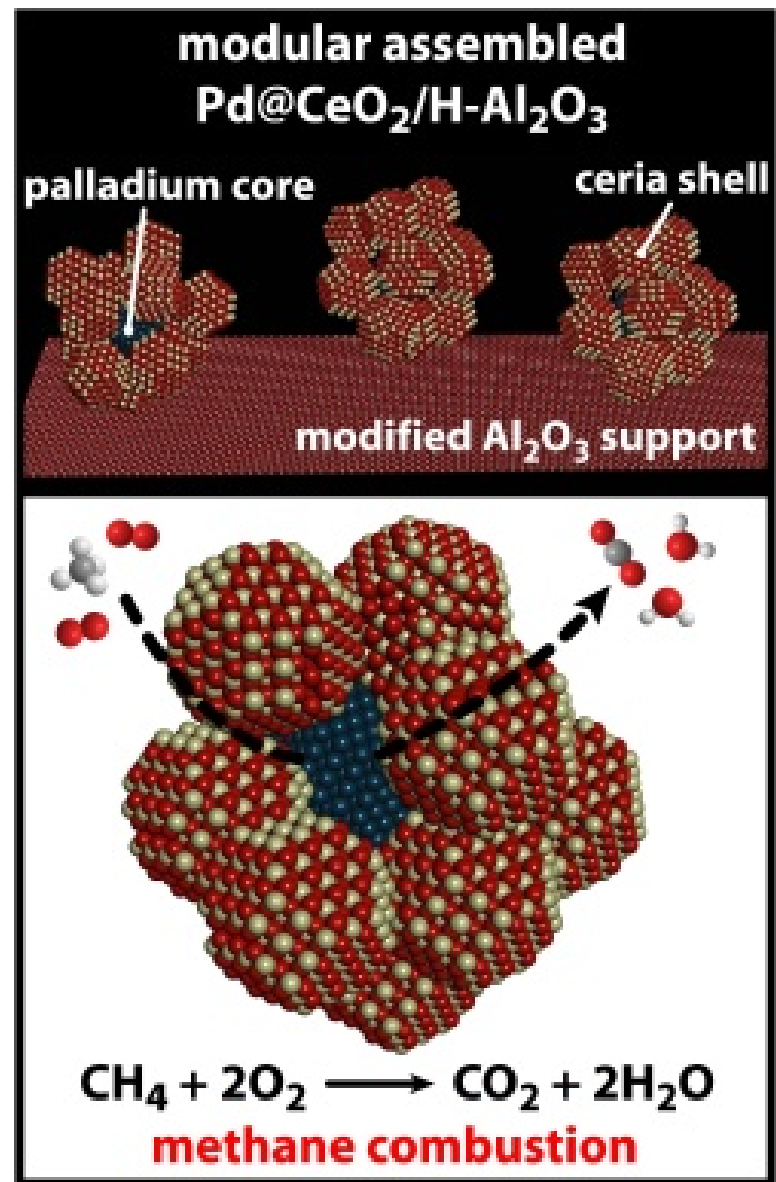


New Catalyst for Efficient, Low Temperature Methane Combustion



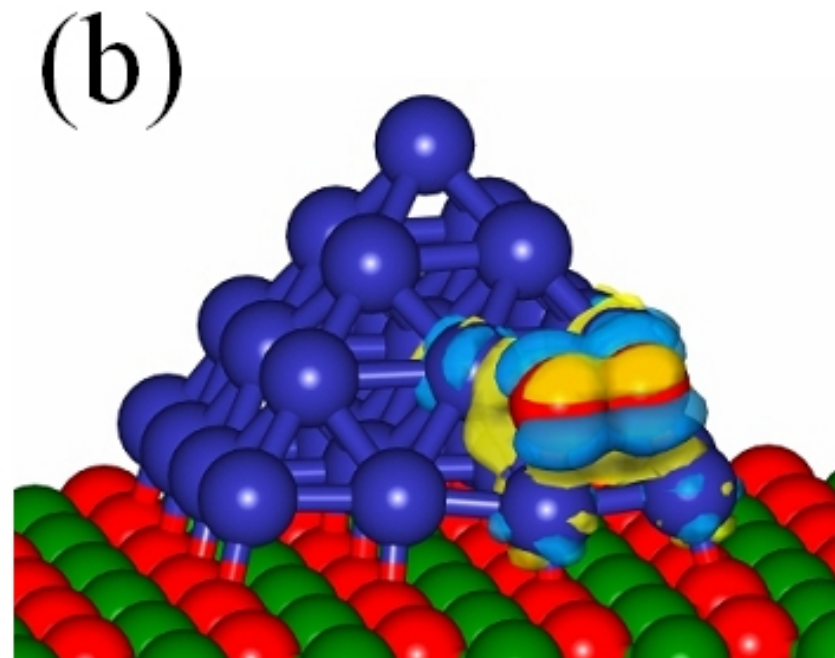
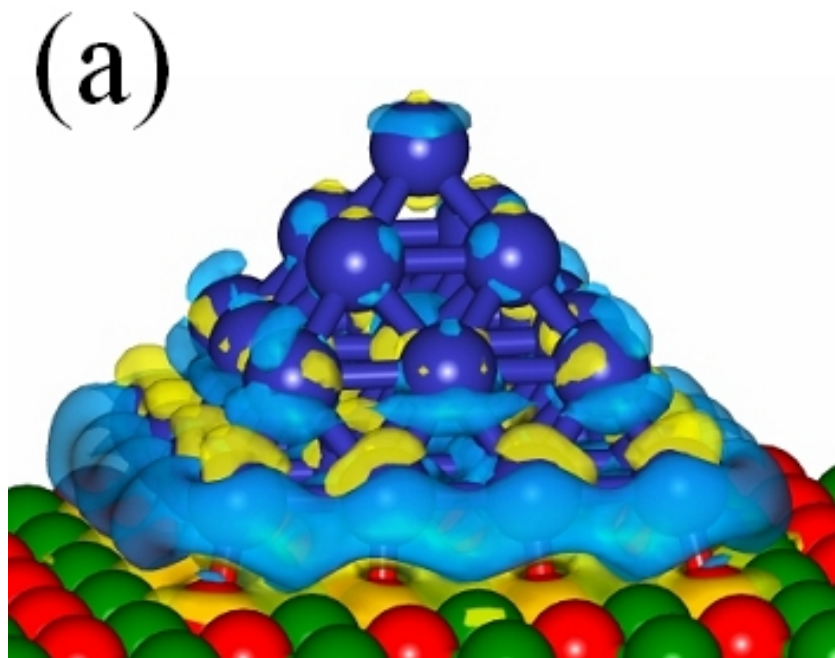
- Novel self-assembled core-shell nanocatalyst
 - 1.8 nm Pd core; Ceria shell
 - PdO forms at interface: active catalyst
- Complete burning of methane at 400°C
- 30 times more efficient than previous catalysts
- Applications:
 - Efficient turbines
 - Steam reforming; WGS
 - Reduce methane in exhausts

Gorte, U Penn (MURI)
Science, 337, 713 (2012)





Catalysis by Pd₃₀ Clusters on MgO



- Charge transfers from MgO substrate to Pd₃₀ cluster at the interface
- Charge (0.25e) goes into anti-bonding orbital of adsorbed O₂
- Reaction stops when extra charge is used up (self-limiting)

Heiz (Munich); Landman (Georgia Tech)
J. Phys. Chem. C 116, 9594 (2012)

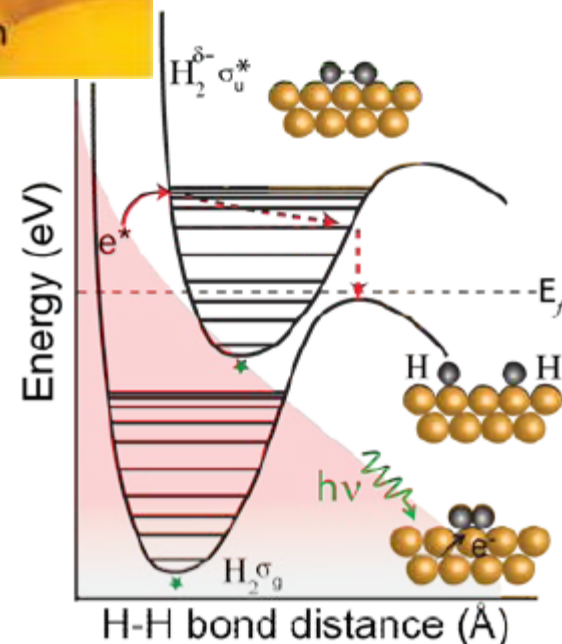
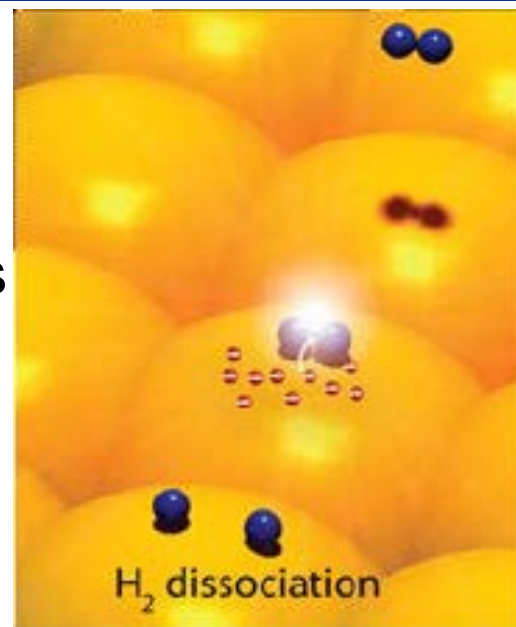


Plasmon-induced Dissociation of H_2 on Gold Nanoparticles



- Visible light excites surface plasmons in Au NP
- Plasmons decay into hot electrons (and holes) with energy below metal work function
- Hot e's transfer to Feshbach resonance of H_2 adsorbed to Au NP triggering dissociation
- New pathway for controlling chemical reactivity on metal catalysts

Halas (Rice U); Carter (Princeton U)

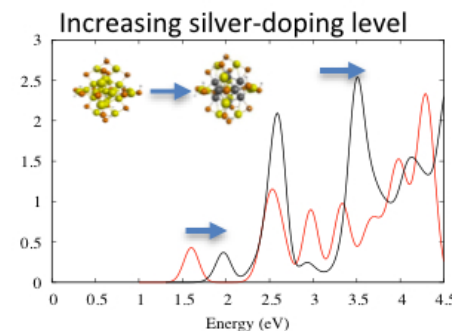
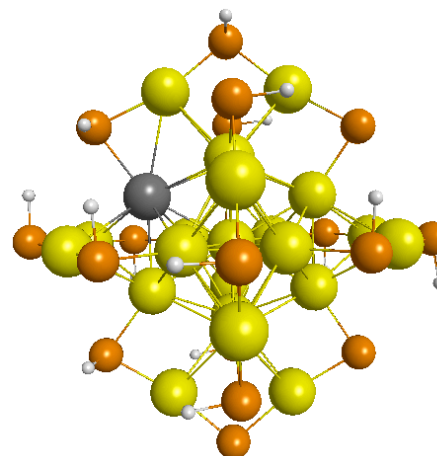
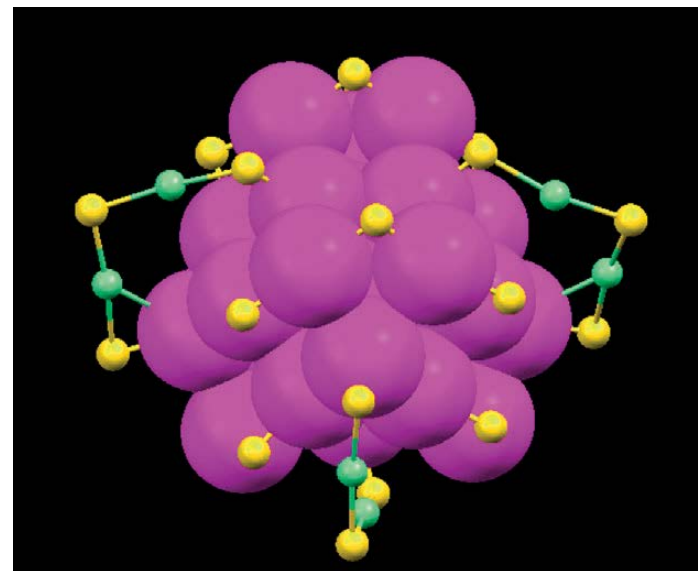




Properties of Gold Thiolate Nanoparticles



- Total Structure and Electronic Properties of the Gold Nanocrystal $\text{Au}_{36}(\text{SR})_{24}$ determined
 - Au_{28} FCC-type tetrahedral core
 - 12 ligands bind to core as bridges
 - 12 ligands bind as “staples”
- Doping with silver atoms provides way to tune optical properties
 - Silver atoms minimize contact with other silver atoms
 - Shifts position and intensity of peaks



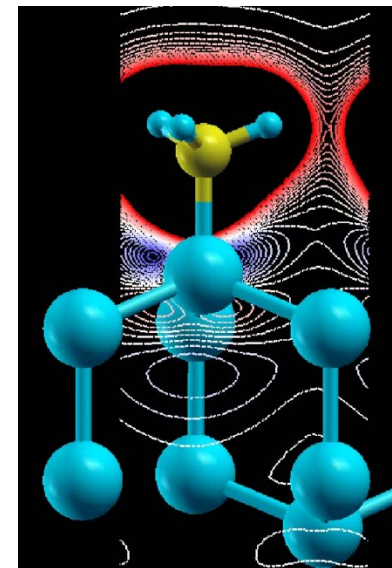
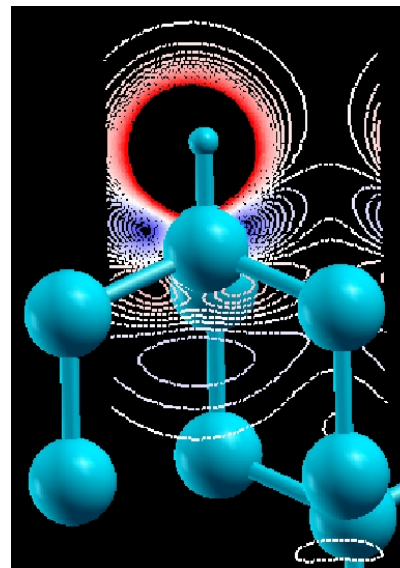
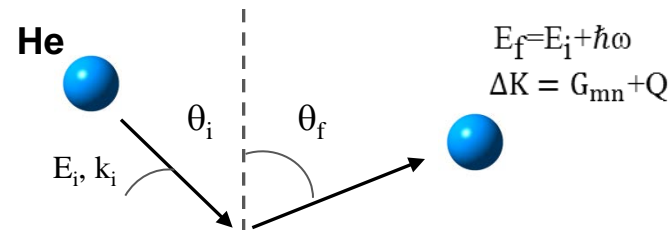
Jin (Carnegie Mellon); Landman (Ga Tech),
Angew. Chem. Int'l. Ed. 51, 1 (2012)



Atom Scattering to Probe Molecular Properties and Interactions



- Elastic and inelastic He atom scattering, STM, & synthesis create new electronic interfaces
- CH₃-Si(111) offers superior chemical and structural stability compared to H-Si(111)
- Air- and electrochemical-stability enables advanced sensors, fuel and solar cells, etc.
- Probes defects (electron traps) and surface librations (bonding, electron-phonon coupling)
- Provides information not attainable with IR/Raman

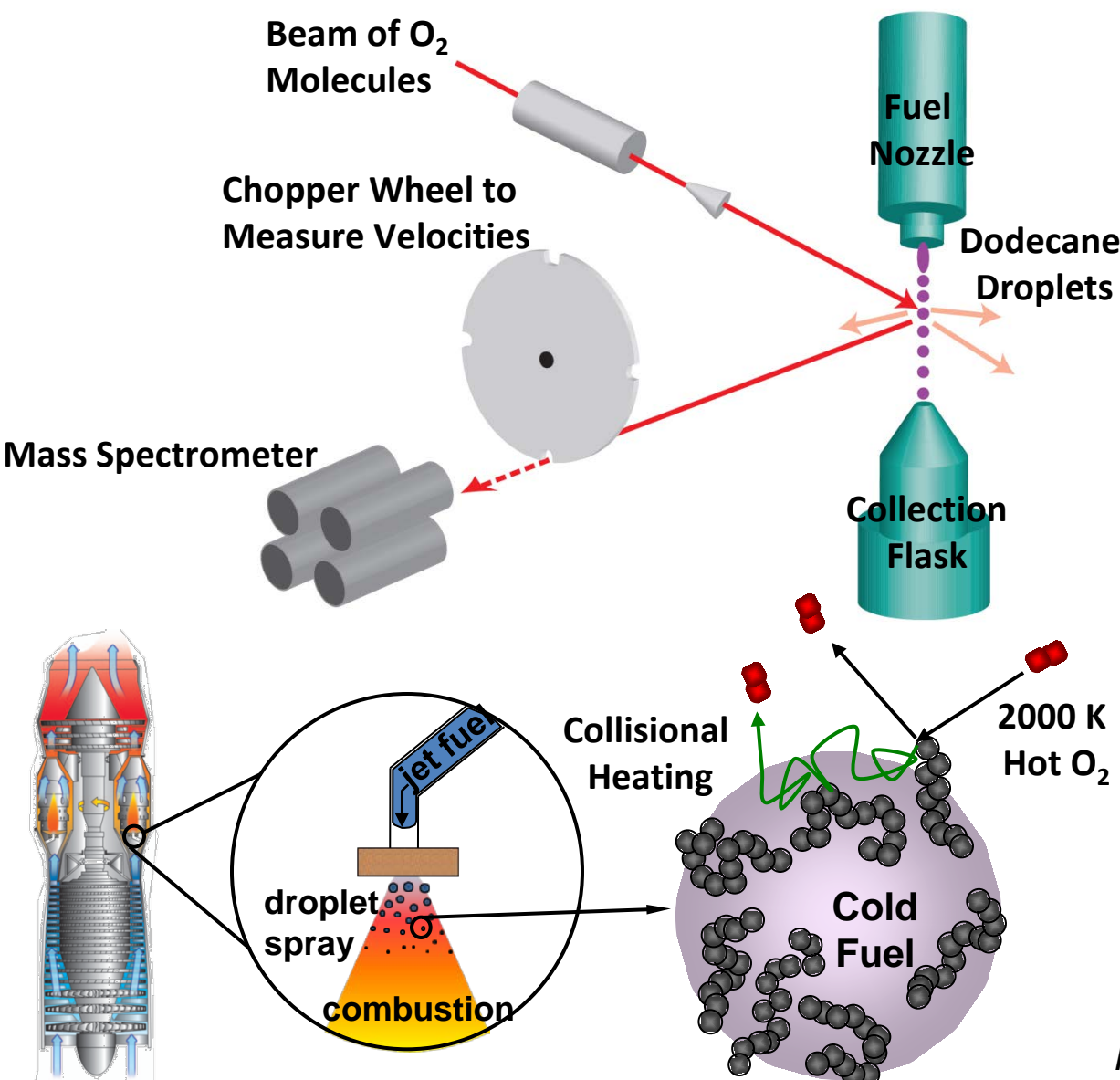


Sibener, (U. Chicago)
J. Chem. Phys. 133, 10470 (2010);
Faraday Discussion 157, 307-323 (2012)

Synthesis: Lewis (Caltech)
Modeling: Benedek (Milan)



Jet Fuel Heating by Hot Gases

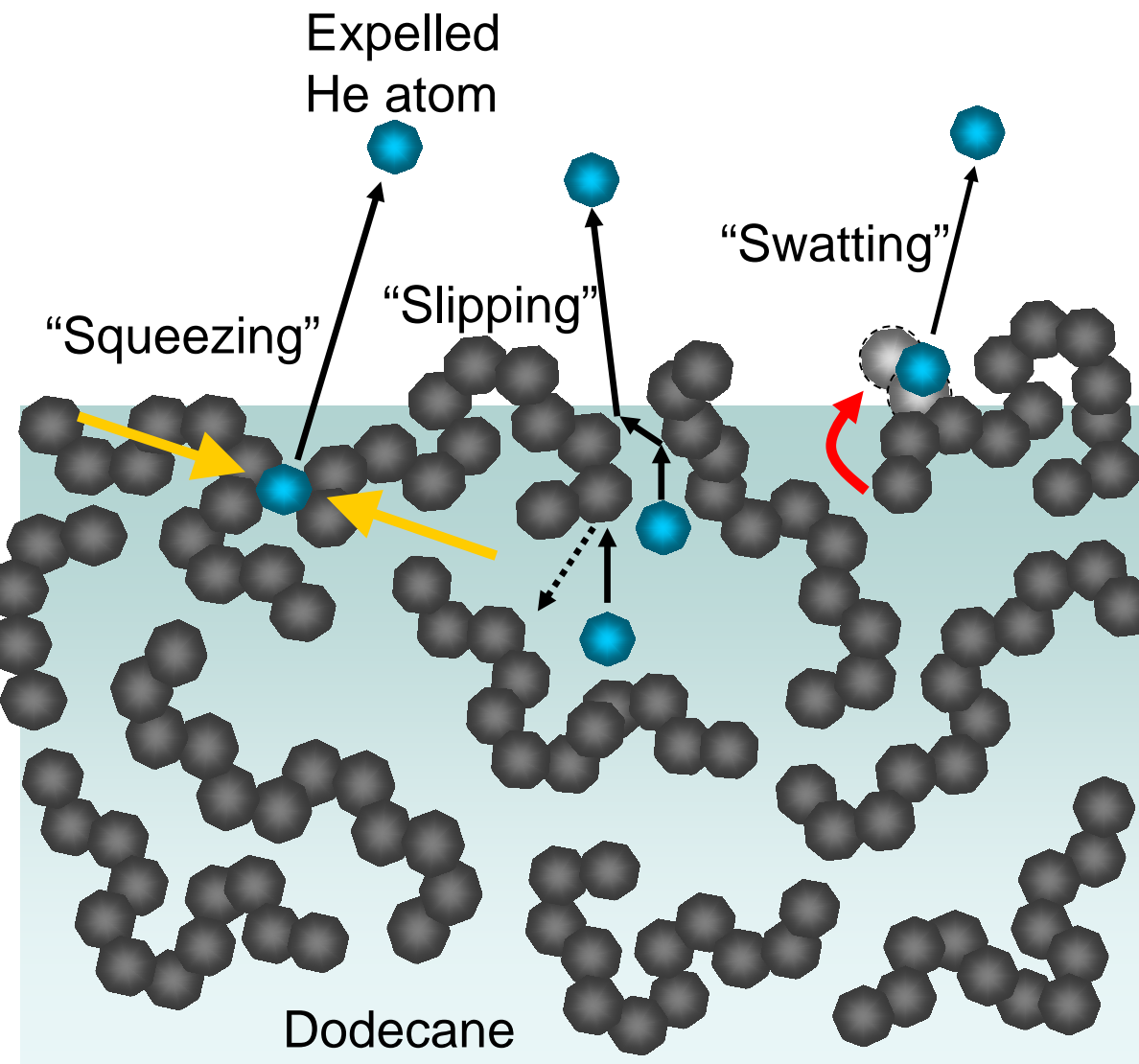


- Understanding fundamental processes in fuel droplets:
 - Heating (energy transfer)
 - Reactions
 - Evaporation
- To model this process, we direct hot O_2 molecules at liquid dodecane in vacuum
 - Enabled by liquid microjets
- O_2 molecules transfer $\sim 2/3$ of their energy upon collision with dodecane.
 - Rapidly heats surface of the droplet
 - Fuel evaporation is not limited by gas-liquid collisions, but by heat transfer in the gas phase.

Nathanson, U. Wisconsin-Madison¹⁸



The Last Leap of a Gas Atom Off the Surface of a Liquid



- He atoms dissolved in dodecane evaporate with kinetic energies much higher than expected.
- Fast He atoms “zoom” ballistically off the surface after moving through fluctuating gaps between molecules.
 - *Squeezing*
 - *Slipping*
 - *Swatting*
- He atoms are expelled from the surface before they can thermally re-equilibrate.
- Learn how different components of jet fuel mix with each other and alter the motions of gas molecules

Nathanson, U. Wisconsin-Madison

Most Read

December 2012
November 2012
October 2012
September 2012
August 2012
July 2012
June 2012
May 2012
April 2012
March 2012
February 2012
January 2012
December 2011
November 2011
October 2011
September 2011
August 2011
July 2011
June 2011
May 2011
April 2011
March 2011
February 2011
January 2011
December 2010
November 2010
October 2010
September 2010
August 2010
July 2010
June 2010

Top 20 Most Read Articles

August 2012

The 20 articles with the most full-text downloads during the month, in descending order.

SELECTED: [Export Citations](#) | [Show/Hide Summaries](#) | [Add to MyArticles](#) | [Email](#)

[Add](#) [View](#)

The potential and flux landscape theory of evolution

Feng Zhang, Li Xu, Kun Zhang, Erkang Wang, and Jin Wang

J. Chem. Phys. **137**, 065102 (2012); <http://dx.doi.org/10.1063/1.4734305> (19 pages)

Online Publication Date: 8 August 2012

Full Text: [Read Online](#) (HTML) | [Download PDF](#)

+ [Show Abstract](#)

+ [Show PACS](#)

Merged-beams for slow molecular collision experiments

Qi Wei, Igor Lyuksyutov, and Dudley Herschbach

J. Chem. Phys. **137**, 054202 (2012); <http://dx.doi.org/10.1063/1.4739315> (9 pages)

Online Publication Date: 6 August 2012

Full Text: [Read Online](#) (HTML) | [Download PDF](#)

+ [Show Abstract](#)

+ [Show PACS](#)

Dissociative electron attachment to C₂F₅ radicals

Sean A. Haughey, Thomas A. Field, Judith Langer, Nicholas S. Shuman, Thomas M. Miller, Jeffrey F. Friedman, and A. A. Viggiano

J. Chem. Phys. **137**, 054310 (2012); <http://dx.doi.org/10.1063/1.4738759> (8 pages)

Online Publication Date: 3 August 2012

Full Text: [Read Online](#) (HTML) | [Download PDF](#)

+ [Show Abstract](#)

Distribution A: Approved for public release; distribution is unlimited



Summary



- Molecular approach key to energy storage and utilization
- Clusters and nanostructures key to exploiting catalysis, plasmonics, and novel materials
- Knowledge of molecular mechanisms is key in developing and optimizing more efficient catalysts
- AFOSR leading the way in applying new tools to understand energy transfer, reaction, and catalytic mechanisms
- Many new areas of opportunity:
 - Exciton dynamics
 - Ultracold chemistry



Challenges in Chemical Dynamics

Molecular Dynamics, Theoretical Chemistry



- **Energetic Materials**

- Energetic ionic liquids
- Energetic nanostructures
- Non-traditional concepts

- **Nanostructures/Catalysis**

- Nanostructures for catalysis
- Photoelectrocatalysis
- Surface-plasmon enhancement

- **Atm/Space Chemistry**

- Upper atmosphere, space
- Signatures & backgrounds
- Ion & plasma processes

- **Lasers and Diagnostics**

- High-Power Gas Lasers
- Novel analytical tools/methods

(Rocket propellants, explosives)

- ⇒ CHNO limit; new approaches
- ⇒ Sensitivity, mechanisms
- ⇒ Safer, penetrating munitions

(Energy, Catalysis, Sensing)

- ⇒ Atomic scale imaging and control
- ⇒ Activity and stability
- ⇒ Size-dependent properties

(Signatures, surveillance)

- ⇒ Hypersonic prop, gas/surf interact.
- ⇒ Rates/mech. of ion-molecule proc's.
- ⇒ Predictive codes, communication

(Infrared lasers, missile defense)

- ⇒ Efficient pumping, energy transfer
- ⇒ Relaxation processes



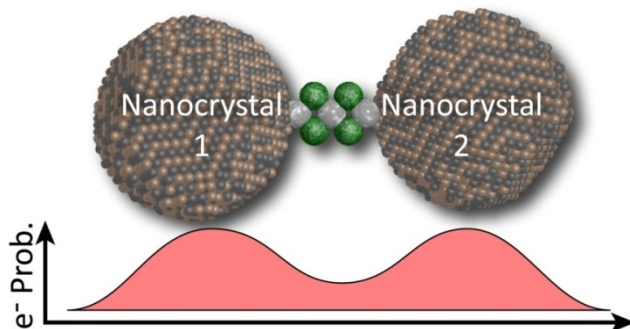
Interactions within AFOSR through BRIs



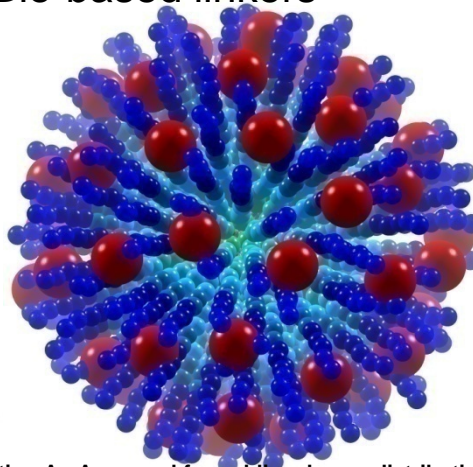
Nanoscale Building Blocks for Novel Materials *with RTD (De Long)*

Use nanoscale structures as building blocks to make novel materials with new properties for energy manipulation

Chemical linkers

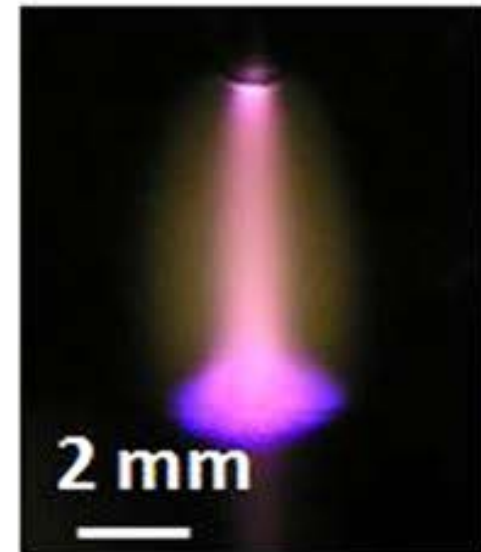


Bio-based linkers



Plasma-Surface Interactions *with RTD (Luginsland)*

Plasma-surface interactions for enabling novel and energy-efficient means of protecting or creating materials



AFOSR Molecular Dynamics and Theoretical Chemistry – Active Grants 2012

Principal Investigator	Institution	Grant Title	Link
Aikens, Christine	KANSAS STATE UNIVERSITY	STRUCTURE AND OPTICAL PROPERTIES OF NOBLE METAL NANOPARTICLES	
Anderson, Scott	UNIVERSITY OF UTAH	(MURI 08) - NANOCATALYSTS IN PROPULSION: MECHANISMS AND OPTIMIZATION	
Bartlett, Rodney	UNIVERSITY OF FLORIDA	MOLECULES AND THEIR INTERACTIONS	
Bergman, Robert	UNIVERSITY OF CALIFORNIA, BERKELEY	(NII) NOVEL CATALYTIC, SYNTHESIS METHODS FOR MAIN GROUP	
Bernskoetter, Wesley	BROWN UNIVERSITY	(YIP 11) Acrylate Formation from CO ₂ and Ethylene by Catalysis	
Bernstein, Elliot	COLORADO STATE UNIVERSITY	STUDY OF HETEROGENEOUS CATALYTIC REACTIONS THROUGH GAS PHASE, NEUTRAL TRANSITION	
Betley, Theodore	HARVARD COLLEGE	(YIP 11) Bifunctional catalysts for CO ₂ reduction	
Bocarsly, Andrew	PRINCETON UNIVERSITY	PHOTOELECTROCHEMICAL CONVERSION OF CARBON DIOXIDE TO ALCOHOLS: FORMATION CARBON-BASED FUELS VIA CARBON-CARBON BOND FORMATION	
Bowen, Kit	JOHNS HOPKINS UNIVERSITY	TOWARD THE DEVELOPMENT OF ALUMINUM CLUSTER-CONTAINING MATERIALS	
Bowers, Michael	UNIVERSITY OF CALIFORNIA SANTA BARBARA	LITIGATED METAL CLUSTERS - STRUCTURES, ENERGY AND REACTIVITY	
Brown, Seth	UNIVERSITY OF NOTRE DAME	(NII) Catalytic activation of nitrogen dioxide for selective synthesis	
Carter, Emily	PRINCETON UNIVERSITY	DESIGNING NEW MATERIALS FOR CONVERTING SOLAR ENERGY TO FUELS VIA QUANTUM MECHANICS	
Castleman, A. Welford	PENNSYLVANIA STATE UNIVERSITY	CLUSTER DYNAMICS: LAYING THE FOUNDATION FOR TAILORING THE DESIGN OF CLUSTER ASSEMBLED NANOSCALE MATERIALS	
Chirik, Paul	PRINCETON UNIVERSITY	(NII) - Synthesis of Fuels and Value-Added Nitrogen-Containing	
Christe, Karl	UNIVERSITY OF SOUTHERN CALIF	POLYNITROGEN CHEMISTRY	
Crim, Fleming	UNIVERSITY OF WISCONSIN	USING VIBRATIONS TO PROBE AND CONTROL PHOTOISOMERIZATION IN LIQUIDS	

Cuk, Tanja	UNIVERSITY OF CALIFORNIA BERKELEY	(YIP 12) In-Situ UV-VIS and IR Spectroscopy of Water Oxidation Catalysts	
Diott, Dana	UNIVERSITY OF ILLINOIS CHAMPAIGN	ULTRAFAST DYNAMICS OF ENERGETIC MATERIALS	
Dukovic, Gordana	UNIVERSITY OF COLORADO	Photophysics and Photochemistry of Nanocrystals with Ultrashort Ligands	
Duncan, Michael	UNIVERSITY OF GEORGIA	Structure, Bonding and Surface Chemistry of Metal Oxide Nanoclusters	
Eichhorn, Bryan	UNIVERSITY OF MARYLAND	(NII) SYNTHESIS AND CHARACTERIZATION OF ALUMINUM BIMETALLIC NANOPARTICLES	
Engel, Gregory	UNIVERSITY OF CHICAGO	(PECASE) - HARNESSING SOLAR POWER NOVEL STRATEGIES FOR RATIONAL DESIGN OF PHOTOCATALYSTS AND PHOTOVOLTAIC MATERIALS	
Fayer, Michael	STANFORD UNIVERSITY	Dynamics and Interactions in Ionic Liquids and Surfaces	
Field, Robert	MASSACHUSETTS INSTITUTE OF TECHNOLOGY	AF-METASTABLE ELECTRONICALLY EXCITED ATOMS AND MOLECULES: EXCITATION TRANSFER IN SLOW COLLISIONS	
George, Steven	UNIVERSITY OF COLORADO	FABRICATION AND PROPERTIES OF ORGANIC-INORGANIC NANOLAMINATES USING MOLECULAR AND ATOMIC LAYER DEPOSITION TECHNIQUE	
Gordon, Mark	IOWA STATE UNIVERSITY	THEORETICAL STUDIES OF GROUP IVA AND GROUP IVB CHEMISTRY	
Halas, Naomi	RICE UNIVERSITY	PLEXICITONICS: COUPLED PLASMON-EXCITON SYSTEMS WITH TAILORABLE PROPERTIES	
Heaven, Michael	EMORY UNIVERSITY	DIODE LASER PUMPED ALKALI VAPOR LASERS WITH EXCIPLEX-ASSISTED ABSORPTION	
Hernandez, Rigoberto	GEORGIA TECH	INCLUDING MOLECULAR STRUCTURE IN THE MULTI-SCALE MODELING OF NON-EQUILIBRIUM FLUID FLOWS	
Jin, Rongchao	CARNEGIE MELLON UNIVERSITY	ON THE EVOLUTION FROM NON-PLASMONIC METAL NANOCCLUSERS TO PLASMONIC NANOCRYSTALS	
Johnson, Mark	YALE UNIVERSITY	Optimizing catalysts for solar fuel production: Spectroscopic characterization of the key reaction intermediates	
Kaiser, Ralf	UNIVERSITY OF HAWAII	(HBCU 09) - UNTANGLING THE ENERGETICS AND DYNAMICS OF ELEMENTARY ATOMIC BORON REACTIONS	
Kanan, Matthew	STANFORD UNIVERSITY	(NII) - Local Electric Field Effects on Rhodium-Porphyrin and NHC-Gold Catalysts	
Khanna, Shiv	VIRGINIA COMMONWEALTH UNIVERSITY	CLUSTER ASSEMBLED MATERIALS FOR NANOENERGETIC APPLICATONS AND CATALYSIS	

Knappenberger, Kenneth	FLORIDA STATE UNIVERSITY	(YIP 10) - MAGNETOPLASMONIC NANOMATERIALS: A ROUTE TO PREDICTIVE PHOTOCATALYTIC, LIGHT-HARVESTING AND FERROFLUIDIC PROPERTIES	
Kubiak, Clifford	UNIVERSITY OF CALIFORNIA SAN DIEGO	(MURI 10) NOVEL CATALYTIC MECHANISMS FOR THE CHEMICAL REDUCTION OF CARBON DIOXIDE TO ENERGY-DENSE LIQUIDS	
Landman, Uzi	GEORGIA TECH	(MURI 08) - ATOMIC-SCALE PRINCIPLES OF COMBUSTION NANOCATALYSIS	
Leone, Stephen	UNIVERSITY OF CALIFORNIA BERKELEY	ULTRAFAST HIGH HARMONIC, SOFT X-RAY LASER PROBING OF MOLECULAR DYNAMICS	
Lineberger, W. Carl	UNIVERSITY OF COLORADO	Thermochemistry and Dynamics of Reactive Species	
Maginn, Edward	UNIVERSITY OF NOTRE DAME	MOLECULAR SIMULATION OF IONIC LIQUIDS: PHYSICAL PROPERTIES, MELTING POINTS, AND DROPLET COLLISION DYNAMICS	
Metiu, Horia	UNIVERSITY OF CALIFORNIA SANTA BARBARA	CATALYSIS BY ATOMIC-SIZED CENTERS: METHANE ACTIVATION FOR PARTIAL OXIDATION AND COMBUSTION	
Michl, Josef	UNIVERSITY OF COLORADO	EXTREMES IN OXIDIZING POWER, ACIDITY, AND BASICITY	
Miller, Thomas	CALIFORNIA INSTITUTE OF TECHNOLOGY	(NII) Exactly Embedded Wavefunction Methods for Characterizing Nitrogen	
Morokuma, Keiji	EMORY UNIVERSITY	THEORETICAL STUDIES OF GAS PHASE ELEMENTARY AND CARBON NANOSTRUCTURE GROWTH REACTIONS	
Nathanson, Gilbert	UNIVERSITY OF WISCONSIN	MOLECULAR BEAM STUDIES OF VOLATILE LIQUIDS AND FUEL SURROGATES USING LIQUID MICROJETS	
Nesbitt, David	UNIVERSITY OF COLORADO	STATE-RESOLVED THERMAL/HYPERTHERMAL COLLISIONAL DYNAMICS OF ATMOSPHERIC SPECIES	
Neumark, Daniel	UNIVERSITY OF CALIFORNIA BERKELEY	New Insights into Catalytic Sites: Spectroscopy of Metal Oxide Clusters	
Pfefferle, Lisa	YALE UNIVERSITY	(MURI 08) - NANOCATALYSTS FOR PROPULSION APPLICATIONS	
Rappe, Andrew	UNIVERSITY OF PENNSYLVANIA	CHEMICAL CONTROL OF EPITAXIAL OXIDE NANOSHELLS AND THEIR ROLE IN WATER SPLITTING	
Rioux, Robert	PENNSYLVANIA STATE UNIVERSITY	Dynamic Chemical and Structural Changes of Heterogeneous Catalysts	

Rogers, Robin	UNIVERSITY OF ALABAMA	DEVELOPING IONIC LIQUID KNOW-HOW FOR THE DESIGN OF MODULAR FUNCTIONALITY, VERSATILE PLATFORMS, AND NEW SYNTHETIC METHODOLOGIES FOR ENERGETIC MATERIALS	
Schatz, George	NORTHWESTERN UNIVERSITY	THEORETICAL STUDIES OF GAS-SURFACE AND GAS-PHASE DYNAMICAL PROCESSES	
Sen, Ayusman	PENNSYLVANIA STATE UNIVERSITY	SELF-MOVING CATALYTIC NANOMOTORS	
Sibener, Steven	UNIVERSITY OF CHICAGO	PRECISION GAS-SURFACE SCATTERING AND IMAGING STUDIES OF COMPLEX INTERFACES	
Subotnik, Joseph	UNIVERSITY OF PENNSYLVANIA	(YIP 11) Nonadiabatic Molecular Dynamics For Electron and Energy Transfer	
Troya, Diego	VIRGINIA TECH	ADVANCING THEORETICAL METHODS TO INVESTIGATE REACTIONS ON ORGANIC SURFACES IN VARIOUS ENERGY REGIMES	
Truhlar, Donald	UNIVERSITY OF MINNESOTA	ORBITAL-DEPENDENT DENSITY FUNCTIONALS FOR CHEMICAL CATALYSIS	
Voth, Gregory	UNIVERSITY OF CHICAGO	UNDERSTANDING THE DRIVING FORCES IN CATALYSIS	
Weiss, Emily	NORTHWESTERN UNIVERSITY	(YIP 10) - ELECTROCHEMICAL REDUCTION OF CO ₂ AT A TiO ₂ ELECTRODE USING QUANTUM DOTS AS MULTI-ELECTRON FUNNELS	
Willets, Katherine	UNIVERSITY OF TEXAS AT AUSTIN	(YIP-09) MAPPING LOCAL SHAPE-DEPENDENT ELECTROMAGNETIC FIELD ENHANCEMENTS IN SINGLE METALLIC NANOPARTICLES USING	

Yarkony, David	JOHNS HOPKINS UNIVERSITY	FROM IGNITION TO PHOTOELECTRON SPECTROSCOPY CONICAL INTERSECTIONS IMPACT THE STUDY OF ENERGETIC MATERIALS	
Zanni, Martin	UNIVERSITY OF WISCONSIN	Dynamics of Photoexcitation at Nanostructured Carbon Interfaces	
Zare, Richard	STANFORD	KINETIC STUDIES OF REACTIONS IN SOLUTION USING FAST MASS SPECTROMETRY	
Zewail, Ahmed	CALIFORNIA INSTITUTE OF TECHNOLOGY	IMAGING UNDER EXTREME CONDITIONS	
Zheng, Junrong	RICE UNIVERSITY	(YIP 11) Chemical Dynamics at Surfaces of Metal Nanomaterials	